Opium Tutorial 2: Aluminum

Goal: The purpose of this tutorial is to introduce the different pseudopotential construction methods and to discuss transferability testing.

Let's calculate the all-electron wavefunctions for aluminum to see what a reasonable cutoff radius could be. We will use the following param file as a starting point:

[Atom] Al 5 100 2.00 - 200 2.00 - 210 6.00 - 300 2.00 - 310 1.00 - [Pseudo] 2 2.5 2.5 kerker [XC] lda

%> ./opium al al.log ae plot wa

A cutoff radius of 2.5 a.u. seems like a very reasonable cutoff for aluminum based on the wavefunction peak positions. We must consider the environment that we wish to use this potential. Specifically, we want to avoid *core overlap* if possible. For example, let's say we are studying aluminum oxide. The smallest bond length in this system is the Al-O bond which can be as small as 3.2 a.u. If we already know that the oxygen cutoff radius is about 1.3 a.u. then we must construct the aluminium pseudopotential to have a cutoff radius of no more than 1.6 a.u. Let's choose 1.6 a.u. and take a look at the approximate cutoff energy:

%> ./opium al al.log ae ps nl ke rpt

KE report

Unlike the previous tutorial, we will concentrate on the output in the report file. Although, it is always a good idea to check the log file for errors and warnings.

 $\%$ cat al.rpt

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Using the Kerker construction method the Al pseudopotential requires a cutoff energy of roughly 18-35 Ry to acheive a convergence error of between 10-100 meV/electron (the effective cutoff energy is the largest of all orbitals).

Let's remake the Al potential using the method of Troullier and Martins [1]. We do this be specifying tm in the param file:

[Atom] Al 5 100 2.00 - 200 2.00 - 210 6.00 - 300 2.00 - 310 1.00 - [Pseudo] 2 1.6 1.6 tm [XC] lda

. .

```
%> ./opium al al.log ae ps nl ke rpt
```
 $\%$ cat al.rpt

. . .

KE report

Using the Troullier-Martins method the kinetic energy convergence behavior is similar to the Kerker method for this atom.

An alternative to the Troullier-Martins and the Kerker method is the Optimized method [2] (sometimes referred to in the literature as $RRKJ$. This method should, at worst, give the same cutoff energy as Troullier-Martins and usually can do better.

The Optimized method requires two additional construction parameters for each valence state. These are specified in the [Optinfo] keyblock:

```
[Atom]
Al
5
100 2.00 -
200 2.00 -
210 6.00 -
300 2.00 -
310 1.00[Pseudo]
2 1.6 1.6
opt
[Optinfo]
3.0 10
3.0 10
[XC]
lda
```
. . .

The first entry in the [Optinfo] keyblock specifies the cutoff wavevector (q_c) for the pseudopotential construction. The square of this value is the desired cutoff energy for this valence state. The PS section of the report file will show the amount of convergence error resulting from this choice. The second parameter is the size of the basis used in the optimization. This value must be at least 4 and can yield better convergence properties if raised to about 8-10. Raising this value beyond 10 is not recommended.

Unlike the Kerker and Troullier-Martins methods, the Optimized approach constrcuts a pseudodopotential which is "optimized" for the choice of q_c . This means that choosing q_c to be too high or too low will result in a sub-optimal potential. Fortunately, there is additional output which can guide the choice of q_c , this is in the PS section of the report file. Let's try a guess of $q_c = 3$ for both states and analyze the PS report output:

%> ./opium al al.log ae ps rpt

%> cat al.rpt

. . .

PS report

====================Optimized pseudopotential method====================

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Using a q_c of 3.0 for both states, we see that the convergence error is about 83 meV/electron for the s state and about 18 meV/electron for the p state. The total convergence error (the sum of the convergence error/electron \times the number of electrons in that state) in this potential is about 184 meV. Let's adjust the q_c to get the total convergence error to be about 100 meV. Clearly, the s state could use some improvement.

After some tests, a q_c of 3.24 $(\sqrt{10.5})$ is found to be acceptable:

PS report

The p state q_c could also be adjusted, but it is not necessary since the p state is already well converged with a q_c of 3.

At this point, let's compare the convergence error results obtained with the Optmized method to that obtained from the other methods. The Optimized Al pseudopotentail with a cutoff energy of 10.5 Ry in the s state has a convergence error of about 38 meV per electron and the p state has roughly 18 meV error per electron. This is a much lower error per electron compared to the other construction methods. In both the Kerker and Troullier-Martins case, at around 10 Ry, there is much more convergence error for both the s and espescially the p states.

Of course one might want the convergence error to be smaller. Again, by adjusting the q_c values, we can directly control the convergence error in the Optimized method. For example, to achieve convergence error of less than 10 meV per electron, we would find that for the s state, a cutoff energy of 13 Ry is needed whereas for the p state, a cutoff energy of about 25 Ry is necessary. Therefore, this potential requires an overall cutoff of 25 Ry. This is much softer than the other methods. For instance, the Kerker potential required 35 Ry for a convergence error of 10 meV per electron and the Troullier-Martins required about 40 Ry.

Now, let's go back to the previous param file and add a section to test the *trasferability* of the potential. This requires us to select a set of test configurations in order to compare the all-electron (AE) to the non-local pseudo- (NL) eigenvalues. Let's try the $+1$, $+2$, and $+2.5$ states:

[Atom] Al 5 100 2.00 - 200 2.00 - 210 6.00 - 300 2.00 - 310 1.00 - [Pseudo] 2 1.6 1.6

[Optinfo] 3.24 10 3.00 10 $[XC]$ lda [Configs] 3 300 1.00 - 310 1.00 - 300 1.00 - 310 0.00 - 300 0.50 - 310 0.00 -

opt

With the [Configs] keyblock specified, the test configurations will be computed. Note that we will rerun the ae ps nl steps, then run the tc step to do run the test configurations and then end with a rpt step to get the report file. Rerunning the ae ps nl steps is actually not necessary if no changes have been made to the param file, however, it doesn't hurt to do this quick step to avoid confusion later.

%> ./opium al al.log ae ps nl tc rpt

The difference between the AE and NL eigenvalues are computed and written to the report file. Also, the total energy difference between each configuration is computed and the AE and NL values of these are subtracted, yield the second difference, sometimes called "delta,delta E".

%> grep AE-NL al.rpt

We see from the eigenvalue and tail norm errors that the transferability is getting worse as Al becomes more ionized. This is because the reference configuration is the neutral atom. This is not a very good choice for an aluminum oxide calculation. A charged reference configuration would probably do much better. Let's try the following param file:

[Atom]

Al 5 100 2.00 - 200 2.00 - 210 6.00 - 300 0.50 - 310 0.10 - [Pseudo] 2 1.6 1.6 opt [Optinfo] 3.24 10 3.00 10 [XC] lda [Configs] 3 300 1.00 - 310 1.00 - 300 1.00 - 310 0.00 - 300 0.50 - 310 0.00 -

This reference configuration can lead to much better transferability near more the positive charged aluminum species:

Another way to qualitative check the transferability of the pseudopotential is to plot the logarithmic derviatives. This can easily be done by adding the [Loginfo] key block to the param file. In the keyblock, you must specify the configuration number you want to consider, "0" means the reference state, "1" means the first test configuration, etc. Also, you must specify the radius and energy range for the calculation. For instance, Let's use the compute the logarithmic derivatives over the range of -3.0 to $+3.0$ Ry at 1.7 a.u. (the radius must be larger than all of the cutoff radii).

[Atom] Al 5 $1002.00 -$ 200 2.00 - $2106.00 -$ 300 0.50 - 310 0.10 - [Pseudo] 2 1.6 1.6 opt [Optinfo] 6.3 10 6.3 10 [XC] lda [Configs] 3 300 1.00 - 310 1.00 - 300 1.00 - 310 0.00 - 300 0.50 - 310 0.00 - [Loginfo] Ω 1.7 -3.0 3.0

%> ./opium al al.log ae ps nl plot logd

Here, the vertical lines mark the location of the eigenvalues for this configuration. The hallmark of a pseudopotential with good transferabilty is to have the all-electron and pseudo logarithmic derivatives agree closely for a region of about 1-2 Ry around the eigenvalues. We can see that this is true for this potential.

To Explore: We just saw how selecting the reference state can really change the transferability properties of the pseudopotential. Using aluminum, investigate how different reference states effect the transferability over a specific range of ionization states. Is the pseudopotential more transferable when reference state is more ionized or neutral? How sensitive is this behavior for different cutoff radii?

^[1] N. Troullier and J. L. Martins, Phys.Rev.B 43 1993 (1991).

^[2] A. M. Rappe, K. M. Rabe, E. Kaxiras, and J. D. Joannopoulos, Phys. Rev.B 41, 1227 (1991).

FIG. 1: Aluminum all-electron wavefunctions

FIG. 2: Aluminum logirthmic derivatives